

CHEM30005 Organic Chemistry IIIB

Credit Points:	12.50												
Level:	3 (Undergraduate)												
Dates & Locations:	2010, Parkville This subject commences in the following study period/s: Year Long, Parkville - Taught on campus. Lectures and tutorials												
Time Commitment:	Contact Hours: Three 1-hour lectures per week for 4 weeks (semester 1); one 1-hour tutorial per week for 4 weeks (semester 1); Three 1-hour lectures per week for 8 weeks and up to eight 1-hour tutorials (semester 2). Total 48 hours. Total Time Commitment: Estimated total time commitment of 120 hours												
Prerequisites:	<p>One of</p> <table border="1"> <thead> <tr> <th>Subject</th> <th>Study Period Commencement:</th> <th>Credit Points:</th> </tr> </thead> <tbody> <tr> <td>CHEM20014 Organic and Physical Chemistry 2</td> <td>Year Long</td> <td>12.50</td> </tr> <tr> <td>CHEM20022 Organic Chemistry 2</td> <td>Year Long</td> <td>12.50</td> </tr> <tr> <td>CHEM20024 Organic and Inorganic Chemistry 2</td> <td>Year Long</td> <td>12.50</td> </tr> </tbody> </table> <p># 610-220 Organic Chemistry (prior to 2009) # 610-221 Organic & Bio-organic Chemistry (prior to 2009)</p> <p>Concurrent enrolment in 610-325 Organic Chemistry Practical III (/view/2010/610-325) is strongly recommended.</p>	Subject	Study Period Commencement:	Credit Points:	CHEM20014 Organic and Physical Chemistry 2	Year Long	12.50	CHEM20022 Organic Chemistry 2	Year Long	12.50	CHEM20024 Organic and Inorganic Chemistry 2	Year Long	12.50
Subject	Study Period Commencement:	Credit Points:											
CHEM20014 Organic and Physical Chemistry 2	Year Long	12.50											
CHEM20022 Organic Chemistry 2	Year Long	12.50											
CHEM20024 Organic and Inorganic Chemistry 2	Year Long	12.50											
Corequisites:	None												
Recommended Background Knowledge:	None												
Non Allowed Subjects:	<p>Credit cannot be gained for this subject and any of</p> <table border="1"> <thead> <tr> <th>Subject</th> <th>Study Period Commencement:</th> <th>Credit Points:</th> </tr> </thead> <tbody> <tr> <td>CHEM30004 Organic Chemistry IIIA</td> <td>Year Long</td> <td>12.50</td> </tr> <tr> <td>CHEM30016 Reactivity and Mechanism</td> <td>Semester 1</td> <td>12.50</td> </tr> <tr> <td>CHEM30015 Advanced Practical Chemistry</td> <td>Semester 1</td> <td>12.50</td> </tr> </tbody> </table> <p>An additional non-allowed subject combination normally exists between this subject and 610-354 Specialised Topics in Chemistry A (/view/2010/610-354) and 610-351 Specialised Topics in Chemistry B (/view/2010/610-351). However enrolment in either 610-354 Specialised Topics in Chemistry A or 610-351 Specialised Topics in Chemistry B (with a restricted choice of topics) and this subject, may be approved by the subject coordinator.</p>	Subject	Study Period Commencement:	Credit Points:	CHEM30004 Organic Chemistry IIIA	Year Long	12.50	CHEM30016 Reactivity and Mechanism	Semester 1	12.50	CHEM30015 Advanced Practical Chemistry	Semester 1	12.50
Subject	Study Period Commencement:	Credit Points:											
CHEM30004 Organic Chemistry IIIA	Year Long	12.50											
CHEM30016 Reactivity and Mechanism	Semester 1	12.50											
CHEM30015 Advanced Practical Chemistry	Semester 1	12.50											
Core Participation Requirements:	It is University policy to take all reasonable steps to minimise the impact of disability upon academic study and reasonable steps will be made to enhance a student's participation in the University's programs. Students who feel their disability may impact upon their participation are encouraged to discuss this with the subject coordinator and the Disability Liaison Unit.												
Coordinator:	Assoc Prof Uta Wille												
Contact:	<p>Director of Third Year Studies</p> <p>Email: third-year-director@chemistry.unimelb.edu.au (mailto:third-year-director@chemistry.unimelb.edu.au)</p>												

Subject Overview:	<p>Upon completion of this subject, students should comprehend the chemical characteristics of various reactive intermediates (carbocations, carbanions and free radicals), and gain an understanding of the principles of orbital-controlled reactions. They should gain knowledge on the methodologies for carbon-carbon bond formation and functional group transformation for the synthesis of organic compounds and the range of agents available to effect these transformations using the various different classes of reactive intermediates. They should comprehend the procedures for determination of the structures of organic compounds by spectroscopic and chemical techniques.</p> <p>Students should also appreciate the importance of rational, critical and independent thought in chemical science and in the understanding of organic chemistry.</p>
Objectives:	Refer to Overview.
Assessment:	Three to four short tests each of duration less than 1 hour conducted on-line during the semester using the learning management system (LMS) for a total of 20%, one 1-hour exam at the end of semester 1 and one 90-min exam at the end of semester 2 (80% for both written exams).
Prescribed Texts:	J McMurry, Organic Chemistry, 6th Ed. Thomson Brooks/Cole, 2004.
Breadth Options:	<p>This subject potentially can be taken as a breadth subject component for the following courses:</p> <ul style="list-style-type: none"> # Bachelor of Arts (https://handbook.unimelb.edu.au/view/2010/B-ARTS) # Bachelor of Commerce (https://handbook.unimelb.edu.au/view/2010/B-COM) # Bachelor of Environments (https://handbook.unimelb.edu.au/view/2010/B-ENVS) # Bachelor of Music (https://handbook.unimelb.edu.au/view/2010/B-MUS) <p>You should visit learn more about breadth subjects (http://breadth.unimelb.edu.au/breadth/info/index.html) and read the breadth requirements for your degree, and should discuss your choice with your student adviser, before deciding on your subjects.</p>
Fees Information:	Subject EFTSL, Level, Discipline & Census Date, http://enrolment.unimelb.edu.au/fees
Generic Skills:	This subject will provide the student with the opportunity to establish and develop the following generic skills: the ability to comprehend complex concepts and effectively communicate this understanding to the scientific community and in a manner accessible to the wider community; the ability to connect and apply the learnt concepts to a broad range of scientific problems beyond the scope of this subject; the ability to think critically and independently; the ability to problem-solving, and the ability to use conceptual models to rationalise observations.
Notes:	This subject is available for science credit to students enrolled in the BSc (pre-2008 degree), BASc or a combined BSc course.
Related Majors/Minors/Specialisations:	Chemistry