

## 610-102 Chemistry 2

<b>Credit Points:</b>	12.50
<b>Level:</b>	1 (Undergraduate)
<b>Dates &amp; Locations:</b>	2009, This subject commences in the following study period/s: Summer Term, - Taught on campus. Semester 2, - Taught on campus. Lectures, practicals, tutorials/workshops, independent learning tasks, computer-aided learning.
<b>Time Commitment:</b>	Contact Hours: Semester 2: 36 one-hour lectures (three per week), 18 hours of practical activities (6 three-hour practicals), 12 one-hour tutorial/workshop sessions, 6 hours of computer aided learning, 8 hours of independent learning tasks. Summer semester: 30 hours of lectures (five per week), 18 hours of practical activities (6 three-hour practicals), 12 one-hour tutorial/workshop sessions, 6 hours of computer aided learning, 8 hours of independent learning tasks. Total Time Commitment: 120 hours total time commitment.
<b>Prerequisites:</b>	One of # <i>Chemistry 1</i> # 610-141 Chemistry A (prior to 2009) # 610-121 Chemistry A (Advanced Studies) (prior to 2009)
<b>Corequisites:</b>	None
<b>Recommended Background Knowledge:</b>	None
<b>Non Allowed Subjects:</b>	
<b>Core Participation Requirements:</b>	It is University policy to take all reasonable steps to minimise the impact of disability upon academic study and reasonable steps will be made to enhance a student's participation in the University's programs. This subject requires all students to actively and safely participate in laboratory activities. Students who feel their disability may impact upon their participation are encouraged to discuss this with the subject coordinator and the Disability Liaison Unit.
<b>Coordinator:</b>	Dr Spencer Williams
<b>Subject Overview:</b>	The subject provides an introduction to organic acids and bases; nucleophilic substitution reactions; elimination reactions; addition reactions; electrophilic aromatic substitution reactions; nucleophilic addition reactions; organic redox reactions; chemical kinetics; elementary quantum mechanics, atomic spectra and atomic structure; redox reactions and electrochemistry; and transition metal and coordination chemistry.
<b>Objectives:</b>	The aim of the subject is to provide students with an understanding of the reactivity of organic molecules; the nature of chemical change; the structure of the atom; electrochemical processes and the structure and reactivity of metal compounds.  In the practical component, students should develop basic laboratory skills (observation, analytical techniques, report writing); oral communication skills; independent learning skills; an appreciation of the health and safety issues associated with the safe handling and disposal of laboratory chemicals.
<b>Assessment:</b>	A 30-minute on-line mid-semester test (5%); ongoing assessment of practical work (20%); a 3-hour written examination in the examination period (75%). Satisfactory completion of practical work is necessary to pass the subject. Independent learning tasks need to be completed in order to pass the subject.
<b>Prescribed Texts:</b>	S S Zumdahl, <i>Chemical Principles</i> 6th edn, Houghton Mifflin, 2008. J McMurry, <i>Organic Chemistry</i> 7th edn, Thomson Brooks/Cole, 2008.
<b>Breadth Options:</b>	This subject potentially can be taken as a breadth subject component for the following courses:

	<p># <b>Bachelor of Arts</b> (<a href="https://handbook.unimelb.edu.au/view/2009/D09">https://handbook.unimelb.edu.au/view/2009/D09</a>)</p> <p># <b>Bachelor of Commerce</b> (<a href="https://handbook.unimelb.edu.au/view/2009/F04">https://handbook.unimelb.edu.au/view/2009/F04</a>)</p> <p># <b>Bachelor of Environments</b> (<a href="https://handbook.unimelb.edu.au/view/2009/A04">https://handbook.unimelb.edu.au/view/2009/A04</a>)</p> <p># <b>Bachelor of Music</b> (<a href="https://handbook.unimelb.edu.au/view/2009/M05">https://handbook.unimelb.edu.au/view/2009/M05</a>)</p> <p>You should visit <b>learn more about breadth subjects</b> (<a href="http://breadth.unimelb.edu.au/breadth/info/index.html">http://breadth.unimelb.edu.au/breadth/info/index.html</a>) and read the breadth requirements for your degree, and should discuss your choice with your student adviser, before deciding on your subjects.</p>
<b>Fees Information:</b>	Subject EFTSL, Level, Discipline & Census Date, <a href="http://enrolment.unimelb.edu.au/fees">http://enrolment.unimelb.edu.au/fees</a>
<b>Generic Skills:</b>	<p>This subject encompasses particular generic skills so that on completion of this subject students should have developed skills relating to:</p> <ul style="list-style-type: none"> <li># the organization of work schedules which permit appropriate preparation time for tutorials, practical classes and examinations;</li> <li># the use of electronic forms of communication;</li> <li># the utilisation of computer-aided learning activities to enhance understanding;</li> <li># the performance of basic manipulations with laboratory equipment;</li> <li># the recording of observations, the analysis of information and the interpretation data within a laboratory setting;</li> <li># accessing information from the library employing both electronic and traditional means;</li> <li># working collaboratively with other students;</li> <li># the use of conceptual models;</li> <li># problem solving; and</li> <li># critical thinking.</li> </ul>
<b>Notes:</b>	<p>Students enrolled in the BSc (both pre-2008 and new degrees), BASc or a combined BSc course will receive science credit for the completion of this subject.</p> <p>Students with a high level of achievement in <i>Fundamentals of Chemistry</i> may be permitted to enrol in this subject in semester 2 upon successful completion of the computer aided learning modules of <i>Chemistry 1</i> during the winter recess.</p> <p>A laboratory coat and safety glasses are required for laboratory activities.</p> <p>It is recommended that students have access to a molecular model kit.</p>
<b>Related Course(s):</b>	Bachelor of Optometry Bachelor of Veterinary Science(PV)
<b>Related Majors/Minors/Specialisations:</b>	First year chemistry